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Chapter 4 - Test Your Understanding - Level 2 - Application/Analysis - Page 65: 4. Answer. C.  $C_2H_4$ . Work Step by Step. You can tell that A and B have no double bonds since they have a number of hydrogens which is two more than twice the number of carbons. Option D has the same number of carbons and hydrogens, showing that it has a triple bond.

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