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Section 1. • matter-

anything in the

universe that has

mass (m), and

occupies space, and

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all matter is made of particles. •

chemistry- is the study of matter and the changes that it undergoes •

molecule- is a collection of atoms held together by chemical bonds in a characteristic pattern and proportion

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2 - Atoms, ions, and

molecules Lab 1 - This

is a Lab report for a

physics experiment

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explanation of an

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1.2 Element - a

fundamental

substance that can't

be chemically

changed or broken

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an important role in developing many industrially [^] manufactured fertilizers, alkalis, acids, salts, dyes, polymers, drugs, soaps, detergents, metal alloys and other inorganic and organic chemicals including new materials contribute in a big way to the

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national economy.

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Some Basic Concepts
of Chemistry Class 11

Notes Chapter 1 ...

1.1 Chemistry in
Context Chemistry
deals with the
composition,
structure, and
properties of matter,
and the ways by
which various forms
of matter may be

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interconverted. Thus, it occupies a central place in the study and practice of science and technology.

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Overview: Phases
of Matter,
Measurements,
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Numbers,
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Baker, Mrs. (Science) /
AP Chemistry
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Sheets. These are
chapter reviews for
the 6th edition
Loudon textbook.

They don't cover all
of the chapters yet,
but at some point in
future they might be
updated. Ch 1 review;
Ch 2 review; Ch 3

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review; Ch 4 review;
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Loudon Review
Sheets - Organic
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(1 point) (b) What is
the identity of the
limiting reactant? (1

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point) (c) What is the identity of the excess reactant? (7 points)

(d) Determine the concentration of each ion remaining in solution after the precipitation is complete. (7 points)

16. Assign oxidation states to each of the atoms in the following compounds. a. GaO_3 .

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b. KBrO_4 . c ...

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