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study of carbon and its compounds. simplest organic compound, composed of hydrogen and carbon. hydrocarbons with single bonds, "ane" ending. hydrocarbons with double bonds, "ene" ending.

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Hydrocarbons Alkanes Study Guide Answers  
Alkanes have only a single bond in its molecular formula. A hydrocarbon molecule is said to be saturated if there are no double or triple bonds. Alkenes have one or more than one double bond. Answer true or false:

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Alkanes are saturated hydrocarbons ...

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study of carbon and its compounds. simplest  
organic compound, composed of hydrogen and  
carbon. hydrocarbons with single bonds, "ane"  
ending. hydrocarbons with double bonds, "ene"  
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Chapter 21: Hydrocarbons  
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Hydrocarbon Compounds Guide Answers

Chemistry Hydrocarbons Study Guide Answers A controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane,

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4% propane, and 2% butane Study Guide:  
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molecules that only contain two elements:  
carbon and hydrogen. Common hydrocarbons  
include methane, propane, and butane.  
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hydrocarbons, such as alkanes, are nonpolar  
molecules. 22.2 Unsaturated Hydrocarbons  
CHAPTER 22 Study Guide - Quia Chapter 22  
Hydrocarbons Study Guide Answers Hydrocarbons  
are organic molecules that only contain two  
elements: carbon and hydrogen. Common  
hydrocarbons include methane,

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hydrocarbons, and straight-chain alkanes. Use

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each of the terms below just once to complete the passage. homologous series straight-chain alkanes If all of the carbon atoms are linked by single covalent bonds and there are no branches, the compounds are called (1) examples. Chapter 21: Hydrocarbons Page 11/25

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but this life, all but the world.

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Study Guide Key Concepts 22.1 Hydrocarbons •  
Because carbon has four valence electrons,  
car-bon atoms always form four covalent  
bonds. • The carbon atoms in an alkane can be  
arranged in a straight chain or in a chain  
that has branches. • Molecules of  
hydrocarbons, such as alkanes, are nonpolar  
molecules. 22.2 Unsaturated Hydrocarbons

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Characteristics: 1. Generally non-polar (won't

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dissolve in water) 2.Soluble in non-polar solvents 3.Non-electrolytes. 4.Low melting points ORGANIC CHEMISTRY Continuation of Organic Chemistry I. Methods of analysis, chemistry of

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study guide, you will do well on the final  
exam at the

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The simplest organic compounds are hydrocarbons. Other organic compounds can be derived from hydrocarbons. Hence in simple words, we can define organic chemistry as the study of hydrocarbons and their derivatives. The Necessity of Separate Branch for Study of Carbon Compounds: Organic chemistry is a branch of chemistry, which studies carbon ...

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