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PROPERTIES 1.5 where T_c , P_c , V_c ,
and Z_c are critical temperature, critical
pressure, critical volume, and critical
compressibility factor, respectively. Thus,
for p-xylene, $T_c =$

$412.3[0.567+0.106(0.106)^2]^{-1} =$
 $623.0\text{K}(661.8\text{ F})(\text{literature value is } 616.2$
 $\text{K}) P_c = 106.16(0.34+1.378)^{-2} =$
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