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contribution to the
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conformation appears to be the: A) entropy increase from the decrease in ordered water molecules forming a solvent shell around it. B) maximum entropy increase from ionic interactions between the ionized amino

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genetics, evolution basis of
biochemistry. It explains
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fact that its carbon: A) has
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set (32) . . . In each case, the single intermediate is the ES complex. (see page 190 or picture from test bank) A) (a) describes a strict "lock and key" model, whereas (b) describes a transition-state

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S-14. 1. Solubility of

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Ethanol in Water Explain why ethanol ($\text{CH}_3\text{CH}_2\text{OH}$) is more soluble in water than is ethane (CH_3CH_3).
Answer Ethanol is polar; ethane is not.

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