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Using the methods developed in class:

Step 1: Multiply both sides by the
partition function $H = \sum_j (E_j + pV_j) e^{-E_j/pV_j}$

V, j . Step 2: Get the temperature
derivative at constant (N, P) (The
conjugate variable to H in this case) ∂H

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1 + H (E+pV)eEjpV.j. = 1 Dr

(Ej+pV)eEjpV. □T kT2kT2 N,P V,j V,j.

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McClemore, the CFO of Ultra Bread,

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Ω is the partition function for the grand canonical ensemble, where T , V , are fixed.

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and $q = \frac{V}{h^3} (2\pi m k T)^{3/2}$ is the partition function
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following: $\Omega = \frac{e^{\beta \mu} q}{h^3} e^{-\beta \epsilon}$ and $q = \frac{V}{h^3} \int e^{-\beta \epsilon} g(\epsilon) d\epsilon$ is the partition function for the grand canonical ensemble, where T , V , are fixed.

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of an ideal monatomic gas from the following: $\Omega = \frac{1}{h^3} \int \exp(-\beta \epsilon) g(\epsilon) d\epsilon$ and $q = \frac{1}{h^3} \int \exp(-\beta \epsilon) g(\epsilon) d\epsilon$ is the partition function for the grand canonical ensemble, where T , V , are fixed. The characteristic potential

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